## $\alpha$ -Chloro- $\alpha$ , $\beta$ -unsaturated esters from diethyl 2,2-dichloro-1ethoxyvinyl phosphate

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Diethyl 2,2-dichloro-1-ethoxyvinyl phosphate obtained by Perkow reaction from ethyl trichloroacetate and triethyl phosphite affords with carbonyl compounds after lithiation  $\alpha$ -chloro- $\alpha$ ,  $\beta$ -unsaturated esters.

Villieras  $^{1)}$  recently described the synthesis of  $\alpha$ -chloro- $\alpha$ ,  $\beta$ -unsaturated esters ( $\underline{2}$ ) by reacting carbonyl compounds with  $\underline{\text{in situ}}$  generated diethyl 1-chloro-1-ethoxycarbonyl-1-lithiomethane phosphonate. We found that the unsaturated esters  $\underline{2}$  can be prepared alternatively by lithiation of diethyl 2,2-dichloro-1-ethoxyvinyl phosphate ( $\underline{1}$ ) and reaction with aldehydes and ketones (Scheme, Table)

 $\underline{\mathbf{1}}$  can be prepared in 83 % yield from ethyl trichloroacetate and triethyl phosphite  $^{2)}$  .

Table:  $\alpha$ -Chloro- $\alpha$ ,  $\beta$ -unsaturated esters from diethyl 2,2-dichloro-1-ethoxyvinyl phosphate.

Carbonyl compound	product <sup>a)</sup>	İ	yield <sup>b)</sup>	<u>E/Z</u> -ratio <sup>C)</sup>
Cyclohexanone	C1 <sup>1)</sup> CO <sub>2</sub> C <sub>2</sub> H <sub>5</sub>	3=	78 %	-
Nonanal <sup>d)</sup>	n-C <sub>8</sub> H <sub>17</sub> -CH=C C1 CO <sub>2</sub> C <sub>2</sub> H <sub>5</sub>	<u>4</u>	55 %	E/Z = 42/58
Benzaldehyde	C <sub>6</sub> H <sub>5</sub> -CH=C Cl <sup>1)</sup>	51	85 %	E/Z = 47/53
2-Methy1-2-propenal	CH <sub>2</sub> =C-CH=C CO <sub>2</sub> C <sub>2</sub> H <sub>5</sub>	<u>6</u>	50 %	E/Z = 34/66
Hydrolysis with 2 N HCl	H <sub>5</sub> C <sub>2</sub> O <sub>2</sub> C-CH-POC <sub>2</sub> H <sub>5</sub> 3)	7	68 %	- and mass spectra

a) The structures 3 - 7 were characterized by their IR-, 1H-NMR- and mass spectra.
 b) Isolated yield calculated on 1. c) The E/Z-ratio was determined by G.L.C. analysis from the crude product. d) Before addition of the carbonyl compound 1 - 2 ml hexamethyl-phosphorous triamide were added to the reaction mixture.

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